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Rietveld refinement of the mixed boracite
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Key indicators: powder X-ray study; $T = 300$ K; mean $\sigma(\text{O}-\text{B}) = 0.014$ Å; disorder in main residue; R factor = 0.018; wR factor = 0.025; data-to-parameter ratio = 22.1.

The structural characterization of the new iron–zinc heptaborate bromide with composition $\text{Fe}_{1.59}\text{Zn}_{1.41}\text{B}_7\text{O}_{13}\text{Br}$, prepared by chemical transport is reported. A rigid-body model with constrained generalized coordinates was defined in order to hold the positions of the B atoms at reasonable interatomic distances that typically would reach unacceptable values because of the weak scattering power of boron. There are three independent sites for the B atoms of which two are tetrahedrally coordinated. The bond-valence sum around the third B atom, located on a threefold rotation axis, was calculated considering two cases of coordination of boron with oxygens: trigonal-planar and tetrahedral. The contribution of the fourth O atom to the bond-valence sum was found to be only 0.06 v.u., indicating the presence of a very weak bond in the right position to have a distorted tetrahedral coordination in favour of the trigonal-planar coordination for the third B atom. X-ray fluorescence (XRF) was used to determinate the Fe/Zn ratio.

Related literature

The method of preparation was based on Schmid (1965). For related structures, see: Mao *et al.* (1991); Dowty & Clark (1972, 1973); Mendoza-Alvarez *et al.* (1985); Schindler & Hawthorne (1998); Knorr *et al.* (2007). For properties and potential applications of boracites, see: Campa-Molina *et al.* (1994, 2002); Dana (1951); Mathews *et al.* (1997); Smart & Moore (1992). For bond-valence parameters for oxides, see: Brese & O'Keeffe (1991).

Experimental

Crystal data

$\text{Fe}_{1.59}\text{Zn}_{1.41}\text{B}_7\text{O}_{13}\text{Br}$
 $M_r = 544.65$
Trigonal, $R\bar{3}c$
 $a = 8.6081$ (1) Å
 $c = 21.0703$ (3) Å
 $V = 1352.12$ (3) Å³
 $Z = 6$

Cu $K\alpha$ radiation
 $T = 300$ K
Specimen shape: irregular
 $20 \times 20 \times 0.2$ mm
Specimen prepared at 1173 K
Particle morphology: irregular, pale pink

Data collection

Bruker D8 Advance diffractometer
Specimen mounting: packed powder sample container
Specimen mounted in reflection mode

Scan method: step
 $2\theta_{\min} = 8.1$, $2\theta_{\max} = 110.0^\circ$
Increment in $2\theta = 0.02^\circ$

Refinement

$R_p = 0.018$
 $R_{wp} = 0.025$
 $R_{exp} = 0.014$
 $R_B = 0.06$
 $S = 1.89$

Profile function: pseudo-Voigt modified by Thompson *et al.* (1987)
397 reflections
18 parameters

Table 1

Selected geometric parameters (Å, °).

| | | | |
|--------------------------|------------|-----------------------|------------|
| Zn–Br | 2.680 (3) | B1–O4 ^{vii} | 1.451 (13) |
| Zn–Br ⁱ | 3.412 (1) | B1–O5 ^{vi} | 1.49 (3) |
| Zn–O2 ⁱⁱⁱ | 2.130 (4) | B2–O1 | 1.566 (3) |
| Zn–O3 ⁱⁱⁱ | 2.081 (7) | B2–O3 ^{viii} | 1.452 (8) |
| Zn–O4 ^{iv} | 2.035 (4) | B2–O4 | 1.463 (18) |
| Zn–O5 ^v | 2.012 (7) | B2–O5 | 1.453 (17) |
| B1–O2 ^{vi} | 1.506 (14) | B3–O2 | 1.397 (14) |
| B1–O3 ^{vii} | 1.48 (2) | B3–O1 | 2.38 (3) |
| O2–B3–O2 ^{viii} | 119.9 (9) | | |

Symmetry codes: (i) $-y + \frac{1}{3}, -x + \frac{2}{3}, z + \frac{1}{3}$; (ii) $-x + y + \frac{1}{3}, y + \frac{2}{3}, z + \frac{1}{3}$; (iii) $-y + \frac{2}{3}, x - y + \frac{1}{3}, z + \frac{1}{3}$; (iv) $x, x - y, z + \frac{1}{3}$; (v) $x - \frac{1}{3}, y + \frac{1}{3}, z + \frac{1}{3}$; (vi) $-y - \frac{1}{3}, -x + \frac{1}{3}, z - \frac{1}{3}$; (vii) $-y, x - y, z$; (viii) $-x + y, -x, z$.

Data collection: *DIFFRAC/AT* (Siemens, 1993); cell refinement: *FULLPROF* (Rodríguez-Carvajal, 2006; Rodríguez & Rodríguez-Carvajal, 1997, a strongly modified version of that described by Wiles & Young, 1981); data reduction: *FULLPROF*; method used to solve structure: coordinates were taken from an isotypic compound (Mao *et al.*, 1991); program(s) used to refine structure: *FULLPROF*; software used to prepare material for publication: *DIAMOND*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BR2119).